

Practice Question Paper -3 2020-21 Class X Science (086)

Time: 3 Hours

Maximum Marks: 80

General Instructions:

- (i) The question paper comprises four sections A, B, C and D. There are 36 questions in the question paper. All questions are compulsory.
- (ii) Section–A question no. 1 to 20 all questions and parts thereof are of one mark each. These questions contain multiple choice questions (MCQs), very short answer questions and assertion - reason type questions. Answers to these should be given in one word or one sentence.
- (iii) Section-B question no. 21 to 26 are short answer type questions, carrying 2 marks each. Answers to these questions should be in the range of 30 to 50 words.
- (iv) Section-C question no. 27 to 33 are short answer type questions, carrying 3 marks each. Answers to these questions should be in the range of 50 to 80 words.

(v) Section-D – question no. - 34 to 36 are long answer type questions carrying 5 marks each. Answers to these questions should be in the range of 80 to 120 words. (vi) There is no overall choice. However, internal choices have been provided in some questions. A student has to attempt only one of the alternatives in such questions.

(vii) Wherever necessary, neat and properly labeled diagrams should be drawn.

	SECTION A	
No.	Questions	Marks
1	Name the gas generally liberated when an acid reacts with a metal.	1
2	15 mL of water and 10 mL of sulphuric acid are to be mixed in a beaker. State the method that should be followed. OR Name the natural source of each of the following acids- (i) Lactic acid (ii) Tartaric acid.	1
3	What is Tyndall Effect?	1
4	How many covalent bonds are there in pentane(C_5H_{12})?	1

5	Give an example of optical phenomena which occurs in nature due to atmospheric refraction.	1
6	Redraw the diagram given below in your answer book and show the direction of the light ray after reflection from the mirror. $ \begin{array}{c} \hline & & \\ \hline \hline & & \\ \hline & & \\ \hline & & \\ \hline & & \\ \hline \hline \hline & & \\ \hline \hline \hline & & \\ \hline \hline \hline \hline$	1
7	For the current carrying solenoid as shown below, draw magnetic field lines and giving reason explain that out of the three points A, B and C at which point the field strength is maximum and at which point it is minimum.	1
8	Name the physical quantities which are indicated by the direction of thumb and forefinger in the Fleming's right hand rule?	1
9	What do the following circuit symbols represent?	1
10	Give one reason why multicellular organisms require special organs for exchange of gases between their body and their environment. :	1
11	Name the site of exchange of material between the blood and surrounding cells. OR Name the component of blood that helps in the formation of blood clot in the event of a cut.	1

12	How much percentage of solar energy is absorbed by the green plants?	1
	OR Name any two abiotic components of an environment.	
13	Write the balanced chemical equation for the process of photosynthesis	1

and the codes a) Bot b) Bot assertic c) A is	 bestion numbers 14, 15 and 16, two statements are given- one labeled Assertion the other labeled Reason (R). Select the correct answer to these questions from (a), (b), (c) and (d) as given below: b) A and R are true, and R is the correct explanation of the assertion. b) A and R are true, but R is not the correct explanation of the assertion. c) A and R are true, but R is not the correct explanation of the assertion. c) A strue, but R is false. c) S false, but R is true. 	
14	Assertion: Each step or level of the food chain forms a trophic level. Reason: The various components of the ecosystem are interdependent.	1
15	Attempt any one from 15(I) and 15(II). (I)Assertion: Variations arising during the process of reproduction cannot be inherited.	1
	Reason: Variations may lead to increased survival of the individual.	
	OR (II)Assertion:In human beings, the sex of the child depends on whether	
	the paternal chromosome is X (for girls) or Y (for boys) . Reason: In mammals primary sex determination is strictly chromosomal and is also influenced by the environment.	
16	Assertion: A chemical reaction must always be balanced. Reason: Mass can neither be created nor destroyed in a chemical reaction.	1
	er Q. No 17 - 20 contain five sub-parts each. You are expected to answer <u>any</u> sub parts in these questions.	<u>v</u>
17	Read the following and answer any four questions from 17 (i) to 17 (v)	1x4
	COVID-19 is a respiratory disease, one that especially reaches into your respiratory tract, which includes your lungs.Now,think of your respiratory tract as an upside-down tree. The trunk is your trachea, or windpipe. It splits into smaller and smaller branches in your lungs. At the end of each branch are tiny air sacs called alveoli.The new coronavirus travels down your airways. The lining can become irritated and inflamed. In some cases, the infection can reach all the way down into your alveoli.	

17(i) (I	 What is the function of alveoli? a. This is where SARS-CoV-2, the virus that causes COVID-19 finally affects. b. This is where plasma, proteins and blood cells escapes in the tissues c. This is where oxygen goes into your blood and carbon dioxide comes out. d. Alveoli carries fat and drains excessive fluid back into the blood. 	
17(iiĝii) d)	Your blood oxygen level indicates how much oxygen your red blood cells are carrying. High blood oxygenation plays an essential role in ensuring that your muscles, brain and other organs receive the energy they need to function properly. That's why it is so important to measure your current level and determine whether it is within an acceptable range. Why is it important to measure your current oxygen level? a. To know how much blood is flowing b. To know how much oxygen is flowing in the blood. c. To know how your brain is working d.To know the heart rate	
17(iii)	 The lungs are the organ most commonly affected by COVID-19. If large parts of the lungs are affected, people struggle to absorb enough oxygen and are admitted to hospital. Another severe effect is acute respiratory distress syndrome (ARDS) – also known as "wet lung". This sees severe inflammation spread quickly throughout the lungs. People who develop this may need mechanical ventilation in an intensive care unit, sometimes for a prolonged period. COVID-19 has a further, unusual effect on the body. Compared to other respiratory viruses, it causes <u>marked clotting in the small blood vessels of the lungs</u> and other organs. How are the lungs affected by COVID-19? A. blood clotting B. Lack of oxygen C. Severe inflammation Now choose the correct option- a. A and B only b. B and C only c. None of these d. A, B and C or either of these 	







19		answer any four questions of pH is that it is a measure from 0 to 14.		1x 4
	0-< 7	7	> 7-14	
	acidic	neutral	basic	
	considered basic. The no 8.5 and for groundwater The pH of pure water (H dioxide in the atmospher 5.2. Because of the associate possible. In general, a water with which could cause prema aesthetic problems such the characteristic <u>"blue-s</u> to treat the problem of lo	a pH < 7 is considered acidi- ormal range for pH in surface systems 6 to 8.5. $I_2(0)$ is 7 at 25°C, but when ex- re this equilibrium results in ciation of pH with atmosphe y recommended that the wat a low pH (< 6.5) could be ac- ature damage to metal piping as a metallic or sour taste, st green" staining of sinks and bw pH water is with the use ould indicate that the water in it can also cause aesthetic pr	e water systems is 6.5 to sposed to the carbon a pH of approximately ric gasses and ter be tested as soon as cidic, soft, and corrosive, g, and have associated taining of laundry, and <u>drains . The primary way</u> of a neutralizer.	
19 (i)	The pH scale indicates- a. concentration of the concentration of the concentration of the dimensional dimensionada dimensionada dimensionada dimensionada dimensionada dimensi	halogen hydroxide hydrogen		
19 (ii)	The substance is acidic a. 0-14 b. 0-7 c. 1-7 d. 7-14	when kits pH lies between t	he range-	
19(iii)	What is the pH value of a. approx. 5.2 b. >8.5 c. 7 d. approx. 6.5	f Rainwater?		

19 (iv)	pH of Hard water can be- a. <7 b. >7 c. >8.5 d. <6.5		
19 (v)	pH of some Common Liquids		
	Vinegar	3.0	
	Coffee	5.0	
	Milk	6.3-6.6	
	Pure Water	7.0	
	Seawater	8.3	
	Among these liquids which is the most acidic a. Vinegar b. Seawater c. Milk d. Pure water	-	

20	Read the following and answer any 4 questions from 20 (i) to 20 (v). When a ray of light falls on any object (polished, smooth, shiny object), light from the object bounces back those rays of light to our eyes.	1x4
20 (i)	What is this phenomenon known as- a. Reflection of light b. Refraction of light c. Dispersion of light d. Scattering of light	
20 (ii)	Image Formation by a Plane Mirror	



20 (v)		
	Two types of reflection are shown in the given picture. Which type of reflection doesn't form an image. a. Regular reflection b. Diffused reflection c. Regular refraction d. Diffused reflection	
	SECTION B	
21	State the role of the following in human digestive system : (i) Digestive enzymes (ii) Hydrochloric acid OR Why do the walls of the trachea not collapse when there is less air in it?	2
22	State any two differences between autotrophic nutrition and heterotrophic nutrition.	2
23	What is a homologous series? Which two of the following organic compounds belong to the same homologous? CH3 ,C2H6, C2H6O, C2H6O2,CH4O OR Give reasons for the following: (a)Diamond has a high melting point. (b)Graphite is a good conductor of electricity.	2

24	(i) Why do calcium found in the form of its compounds and gold in its free state?(ii) Name one lustrous non-metal.	2
25	Give reasons:(i) The extent of deviation of a ray of light on passing through a glass prism depends on its colour.(ii) Lights of red colour are used for danger signals.	2
26	How much current will an electric bulb draw from 220 V source if the resistance of the bulb is 1200Ω ? If in place of bulb, a heater of resistance 100 Ω is connected to the sources, calculate the current drawn by it.	2

	Section C	
27	 A blue colour flower plant denoted by BB is crossbred with that of white colour flower plant denoted by bb. (a) State the colour of flowers you would expect in their F1 generation plants. (b) What must be the percentage of white flower plants in F2 generation if flowers of F1 plants are self-pollinated? (c) State the expected ratio of the genotypes BB and Bb in the F2 progeny. OR State the meaning of inherited traits and acquired traits. Which of the two is not passed on to the next generation? Explain with the help of an example. 	3
28	Why bacteria and fungi are called decomposers? List any two advantages of decomposers to the environment.	3
29	 (a) Draw a diagram to show the nutrition in Amoeba and label the parts used for this purpose. Mention any other purpose served by this part other than nutrition. (b) Name the glands associated with digestion of starch in the human digestive tract and mention their role. (c) How is required pH maintained in the stomach and small intestine? 	3

20			3
30	A Name the type of chemical reaction represented by the following equation:		
	(i) $C_{2}O + HO \rightarrow C_{2}$	OH)	
	(i) CaO + H ₂ O \longrightarrow Ca((ii) 3BaCl ₂ + Al ₂ (SO ₄) ₃ \longrightarrow		
	(<i>iii</i>) 2FeSO ₄ $\xrightarrow{\text{heat}}$ Fe ₂ O ₃	$+ 50_2 + 50_3$	
31	The position of three elements A, B and below:	d C in the Periodic Table is shown	3
	Group 16	Group 17	
	-	-	
	-	A	
	В	С	
	Giving reasons, explain the following:(a) Element A is a non-metal.(b) Element B has a larger atomic size(c) Element C has a valency of 1	than element C.	
32	State three reasons for the following fa (i) Sulphur is a non-metal (ii)Magnesium is a metal One of the reasons must be supported w		3
33	The formation of rainbow in the sky has which phenomenon of the light? Explandiagram.	11	3
	Section	D	
34	State reason for the following statements	s:	5
	(i) Tap water conducts electricity wherea		
	(ii) Dry hydrogen chloride gas does not	turn blue litmus red whereas dilute	
	hydrochloric acid does.(iii) During the summer season, a milkm of baking soda to fresh milk.	aan usually adds a very small amount	
	I VI DONTER MARCHULLANT HURS		
	(iv) For a dilution of acid, acid is added	into water and not water into acid.	

	OR	
	Equal length of magnesium ribbon are taken in two test tubes 'A' and 'B $\$ H2SO4 is added to test tube 'A' and H2CO3 in the test tube 'B' in equal amounts:	
	(a) Identify the test tube showing vigorous reaction.	
	(b) Give reason to support your answer.	
	(c) Name the gas liberated in both the tubes. How will you prove its liberation?	
	(d) Write chemical equations for both reactions.	
	(e) Out of the two acids taken above which one will have	
	(i) lower pH value	
	(ii) lower H+ concentration respectively.	
35	(a) Name the parts labelled A, B, C, D and E.	5
	 (b) Where do the following functions occur? (i) Production of an egg (ii) Fertilisation (iii) Implantation of zygote. 	
	(c) What happens to the lining of uterus:(i) before release of a fertilised egg?(ii) if no fertilisation occurs?	

